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2003 Mass Number

Find the number of
neutrons in Lithium 7.

$(\text{Mass \#}) - (\text{Atomic \#})$
 $= (\# \text{ of neutrons})$

Mass
Number: number of
protons and neutrons -

Atomic Number:

number of protons

Number of Neutrons

Ex. Tritium is Hydrogen
with a mass number of

3. Find the number of

neutrons. $(\text{Mass \#}) -$

$(\text{Atomic \#}) = (\# \text{ of}$

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Chemistry
(neutrons)

Answers

**Period: Isotopes;
Molecular Notation;
Electron Orbits
Isotopes**

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